

1   **Q.    Please explain what further environmental benefits and improved**  
2       **emissions would arise by use of a .5% as opposed to a 1% sulphur fuel**  
3       **oil. Would the use of .5% sulphur fuel oil be a lesser cost alternative to**  
4       **retrofitting the Holyrood facility?**

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6  
7   **A.**    The formula specified by the Department of Environment and Conservation  
8           for calculation of annual emissions of sulphur dioxide at the Holyrood facility  
9           is:

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11       
$$\text{SO}_2 \text{ (kg)} = (19.579 * S * \text{barrels of fuel consumed} * 158.9873 * \text{specific}$$
  
12       
$$\text{gravity}) / 1000$$

13  
14       Where

15  
16       19.579 = emission factor, kg/1000 litres

17       158.9873 = conversion factor, litres/barrel

18       Specific Gravity =  $141.5 / (131.5 + \text{API})$

19       S = percent sulphur content as a whole number

20  
21       As S is one of the three variables in this formula, the others being the barrels  
22       of fuel consumed and the specific gravity, use of a .5% as opposed to a 1%  
23       sulphur fuel oil will have a direct effect on the volume of SO<sub>2</sub> emissions  
24       resulting.

25  
26       US Environmental Protection Agency document AP 42, Fifth Edition,  
27       *Compilation of Air Pollutant Emission Factors, Volume 1: Stationary Point*  
28       *and Area Sources* states:

1  
2 “Sulphur oxides (SO<sub>x</sub>) emissions are generated during oil combustion from  
3 the oxidation of sulphur contained in the fuel. The emissions of SO<sub>x</sub> from  
4 conventional combustion systems are predominantly in the form of SO<sub>2</sub>.  
5 Uncontrolled SO<sub>x</sub> emissions are almost entirely dependent on the sulphur  
6 content of the fuel and are not affected by boiler size, burner design, or  
7 grade of fuel being fired. On average, more than 95 percent of the fuel  
8 sulphur is converted to SO<sub>2</sub>, about 1 to 5 percent is further oxidized to  
9 sulphur trioxide(SO<sub>3</sub>), and 1 to 3 percent is emitted as sulphate particulate.”

10  
11 Therefore, the emission rate for SO<sub>2</sub> from the Holyrood facility can be  
12 expected to be directly proportional to the sulphur content of the fuel. In this  
13 case, there would be a 50% reduction.

14  
15 US EPA AP 42 also identifies a link between the sulphur content of No. 6 oil  
16 and particulate matter emissions. AP 42 states:

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18 “Filterable particulate matter emissions depend predominantly on the grade  
19 of fuel fired. Combustion of lighter distillate oils results in significantly lower  
20 PM formation than does combustion of heavier residual oils. Among residual  
21 oils, firing of No. 4 or No. 5 oil produces less PM than does the firing of  
22 heavier No. 6 oil.

23  
24 In general, filterable PM emissions depend on the completeness of  
25 combustion as well as on the oil ash content. The PM emitted by distillate  
26 oil-fired boilers primarily comprises carbonaceous particles resulting from  
27 incomplete combustion of oil and is not correlated to the ash or sulphur  
28 content of the oil. However, PM emissions from residual oil burning are  
29 related to the oil sulphur content. This is because low-sulphur No. 6 oil,

1        either from naturally low-sulphur crude or desulphurized by one of several  
2        processes, exhibits substantially lower viscosity and reduced asphaltene,  
3        ash and sulphur content, which results in better atomization and more  
4        complete combustion.”

5  
6        Therefore, use of a .5% as opposed to a 1% sulphur fuel oil would result in  
7        reduced particulate matter emissions. The formula recommended in US  
8        EPA AP 42 for calculating particulate emissions rates from uncontrolled  
9        industrial boilers firing No. 6 oil based on sulphur content of the fuel is:

10  
11       8.34(1.12(S) + 0.37) (in lb/10<sup>3</sup> gal) where S is the weight percent sulphur  
12       content of the fuel.

13  
14       Using this formula, the projected particulate emission rate using 1% sulphur  
15       fuel would be 12.43 lb/10<sup>3</sup> gallons, and that using 0.5% sulphur fuel would be  
16       7.76 lb/10<sup>3</sup> gallons, or a 38% reduction.

17  
18       The use of 0.5% sulphur fuel oil would be a lesser cost alternative to  
19       retrofitting the Holyrood facility with a flue gas desulphurization / electrostatic  
20       precipitator (FGD/ESP) to achieve the same reduction in emissions.

21  
22       As noted in CA 14 NLH(c), the price premium for low sulphur fuel oil such that  
23       the economic/financial advantage of using 1% S fuel oil option over FGD/ESP  
24       would be eliminated would be \$9.90 per barrel. In CA 1, the forecast price  
25       premium for 0.5% S fuel oil over 2.0% fuel oil is below \$9.90 except for the  
26       last three years of the period in question. As well, as indicated on pages 4-5  
27       and 4-6 of the *Cost Analysis of FGD vs Low Sulphur Fuel Oil – Air Emissions*  
28       *Controls Assessment – Holyrood Thermal Generating Station Final Report*,  
29       SGE Acres, February 2004, in order to achieve a reduction in SO<sub>2</sub> emissions

1        equivalent to using 0.5% S fuel oil, the capital cost of the FGD would also  
2        increase significantly, compared to the capital cost of the FGD in order to  
3        achieve a reduction in SO<sub>2</sub> emissions equivalent to using 1.0% S fuel oil.